



Effect of SnO₂ Doping on the Performance of High Voltage ZnO Varistors

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Abstract. The effect of the SnO₂ doping on the structure and the electrical performance of ZnO varistors with high voltage was systematically studied. When the doping amount of SnO₂ was small, the main effect was to promote the growth of grains. As the doping amount increased, SnO₂ led to the formation of the Zn₂SnO₄ spinel, inhibiting the growth of ZnO grains. The donor concentration, the boundary barrier and the surface state density decreased first and then increased, respectively. By proper SnO₂ doping, the key performance parameters of the high voltage varistors were improved, where the breakdown voltage gradient was increased from 193.0 to 208.5 V · mm⁻¹, the voltage ratio was reduced from 1.74 to 1.73, and the 2 ms square waveform impulse energy withstanding capacity was increased from 200 to 250 A.

Keywords: SnO₂ Doping · ZnO Varistors · Energy withstanding Capacity · Doping Amount

1 Introduction

Zinc oxide varistor is a kind of semiconductor ceramic element made of ZnO as the main material and various additives [1–7]. ZnO varistors have been used for the protection of power transmission and distribution because of their excellent voltage dependent property, fast response time and strong energy withstanding capacity [8–12].

Doping can improve the nonlinear voltammetry characteristics of ZnO varistors, increase or decrease grain resistance, promote or inhibit the growth of grains, thus affecting the properties of ZnO varistors [13–17]. Therefore, doping is an important way to improve the performance of ZnO varistors. As a flux, SnO₂ doping can reduce the phase transition temperature of Bi₂O₃ in the sintering process, so as to reduce the potential gradient of the varistors, which is mostly used in low gradient varistors or high energy varistors. Zou Qingwen [18] doped SnO₂ into ZnO varistors and found that the total amount of additives in the new formulation system was significantly decreased, and the protection ability was stronger. Zhijun Xu et. al. [19] investigated the effects of SnO₂

on the ZnO varistor ceramics. They found that a small amount of SnO₂ caused a high nonlinear coefficient, lower leakage current, as well as noticeably higher breakdown voltage gradient. There have been several reports on the effect of SnO₂ doping on low voltage ZnO varistors, but there are few reports on the influence of SnO₂ doping on the performance, especially on the large energy withstanding capacity of the varistors with high voltage used in power transmission and distribution systems.

Herein, the effect of SnO₂ doping on the properties of the high voltage ZnO varistors was studied in detail. The influence of the doped SnO₂ on the donor concentration, surface density of states, and grain boundary barrier of ZnO varistors was estimated. With appropriate doping amount, the potential gradient, residual voltage ratio and large energy withstanding capacity of the ZnO varistors were enhanced compared to those of the varistors without SnO₂ doping.

2 Experimental Procedure

The ZnO varistor is composed of ZnO (95 mol%) as main material and the other oxides (5 mol%) including Cr₂O₃, Bi₂O₃, Co₂O₃, MnCO₃, Sb₂O₃, NiO and SnO₂ (*x* mol%, *x* = 0, 0.1, 0.2, 0.3, 0.4, and 0.5) as additives. The samples were named as D0, D1 - 5 according to the content of SnO₂. The ZnO powder and the additive powder were ball milled for 24 h. Centrifugal spray dryer (LZG-5, Wuxi Fenghua Drying Equipment Co., LTD, China) was used for spray granulation of the total slurry. A powder hydraulic press (Y79-25, Shanghai Huci Electric Appliance Development Co., LTD, China) was used for pressing. The green discs were heated to 500 °C to remove organics, then were sintered at 1200 °C to obtain the varistors. Both sides of the sintered samples were coated with Al electrodes.

The crystal phase of the samples was analyzed by an X-ray diffractometer. The morphology of the samples was characterized by a scanning electron microscope. Breakdown voltage $U_{1\text{mA}}$, leakage current I_L , nonlinear coefficient α , and E-J characteristic curve of the samples were measured by a DC varistor parameter tester. The C-V characteristic parameters were tested by a precision impedance analyzer. The donor concentration N_d , the height of the Schottky barrier φ and the surface density N_s were calculated from the C-V plots according to the equation $(\frac{1}{C_b} - \frac{1}{2C_0})^2 = \frac{2(\varphi - V_b)}{q\epsilon_0\epsilon_r N_d}$, where C_b is the grain boundary capacitance per unit area, C_0 is the C_b value when the applied bias is 0, and V_b is the bias of the individual grain boundary, φ is the barrier height, q is the charge of an electron, ϵ_0 and ϵ_r are the vacuum permittivity and the ZnO relative permittivity, respectively [20, 21]. The aging performance of the varistors was tested by a varistor accelerated aging tester under an AC voltage of $85\%U_{1\text{mA}}/\sqrt{2}$ at 135 °C for 96 h. The aging performance was evaluated using the aging coefficient K_{ct} ($K_{ct} = P_{96}/P_1$). P_{96} is the power consumption after 96 h, P_1 is the power consumption after 1 h. A K_{ct} value less than 1 indicates a good aging performance.

3 Results and Discussion

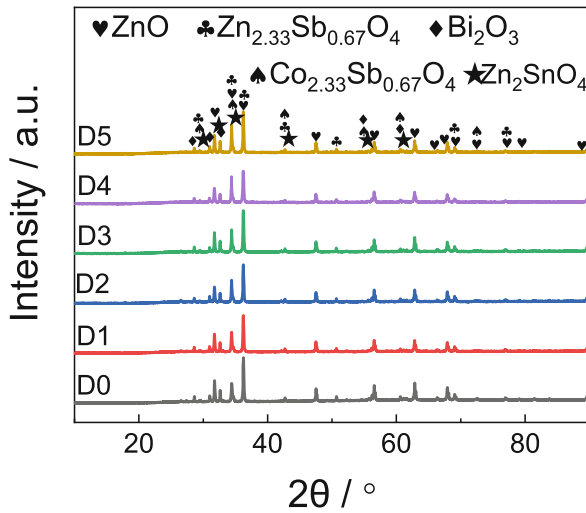


Fig. 1. XRD patterns of ZnO varistors of D0 - D5.

Figure 1 displays the XRD patterns of the ZnO varistors with different SnO₂ doping content. The main structure is composed of hexagonal crystalline ZnO phase, bismuth-rich phase δ -Bi₂O₃ and spinel phases Zn_{2.33}Sb_{0.67}O₄ and Zn₂SnO₄. The diffraction patterns of all samples are similar, indicating that SnO₂ does not have a visible effect on the crystal phase of ZnO varistors, which may be because the content of SnO₂ is small.

Figure 2 shows the SEM images of the ZnO varistors with different contents of SnO₂. The average grain size of the sample without SnO₂ was 8.16 μm . With the increase of SnO₂, the grain size increases to 8.98 μm , and then decreases to 7.24 μm . When the doping amount of SnO₂ is small, the main effect is to reduce the phase transition temperature of Bi₂O₃, make the liquid phase form at lower temperature, promoting the grain growth [18]. The grain size distribution of samples D1 - D5 gradually changed from uneven to uniform, and the average grain size gradually decreased, because as the SnO₂ content increased, the inhibiting effect of the spinel on the grain growth was enhanced.

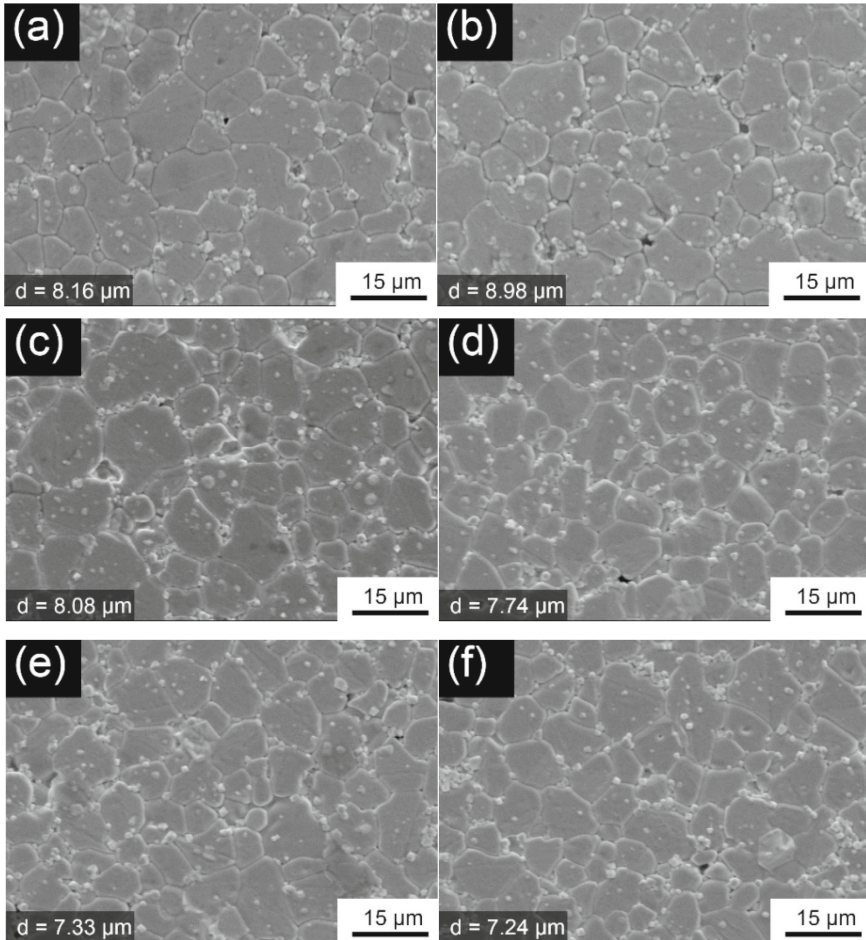


Fig. 2. SEM images of the ZnO varistors of (a) D0, (b) D1, (c) D2, (d) D3, (e) D4, and (f) D5.

Figure 3 (a) shows the potential gradient curves of the ZnO varistors with different amounts of SnO₂ doping. As the SnO₂ amount increases, the potential gradient decreased first from 193.0 to 190.5 V·mm⁻¹, then increased to 218.9 V·mm⁻¹. The decrease of the potential gradient is due to the appearance of some voids in the grain and the increase of the grain size. As SnO₂ increases gradually, the grain size decreases and the distribution is uniform, so the potential gradient for D2 - D5 is increasing.

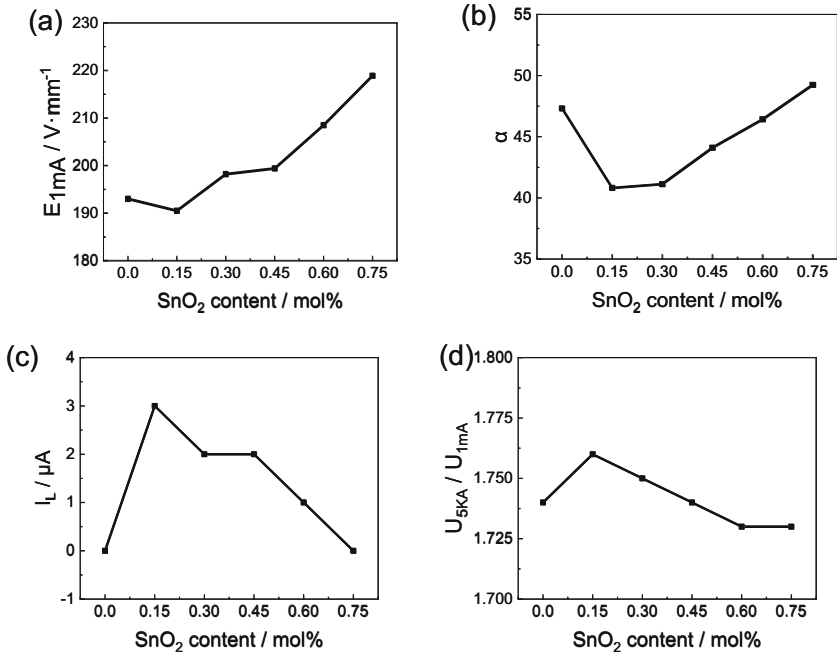


Fig. 3. Properties of ZnO varistors including breakdown voltage gradient (a), nonlinear coefficient (b), leakage current (c) and residual voltage ratio (d) as functions of the SnO₂ content.

Figure 3 (b) displays the variation curve of the nonlinear coefficient of the varistors with different SnO₂ doping contents. With the increasing of the SnO₂ content, the non-linearity decreases first and then increases. The nonlinear decrease of D1 is due to the lack of spinel particles and uneven distribution. With the increase of the SnO₂ content, the Zn₂SnO₄ spinel was formed on the grain boundary. Spinel particles hindered ion migration and increased the barrier height of the grain boundary.

Figure 3 (c) shows the leakage current curve of the ZnO varistors with different SnO₂ doping contents. With the increase of the SnO₂ content, the leakage current firstly increased and then decreased. According to the formula $J = J_0 \exp\left(-v\varphi_B^{\frac{3}{2}}/E\right)$, where J is the current density, J_0 is the current density when the bias is 0, J is inversely proportional to φ , that is, inversely proportional to the nonlinearity [22].

Figure 3 (d) displays the variation of the residual voltage ratio of the ZnO varistors as a function of the SnO₂ doping amount. As the SnO₂ doping amount increases, the residual voltage ratio increased first and then decreased, because E_{5kA} is inversely proportional to E_{1mA} , according to the formula $K = E_{5kA}/E_{1mA}$.

Table 1 shows the detail parameter values of the electrical properties shown in Fig. 3.

Table 1. Electrical properties of the ZnO varistors with different amounts of SnO₂ doping.

Sample	Content (mol%)	E_{1mA} (V/mm ⁻¹)	α	I_L (μ A)	K (U_{5kA}/U_{1mA})
D0	0	193.0	44.31	0	1.74
D1	0.15	190.5	39.81	3	1.76
D2	0.30	198.2	40.12	2	1.75
D3	0.45	199.4	41.10	2	1.74
D4	0.60	208.5	43.42	1	1.73
D5	0.75	218.9	45.24	0	1.73

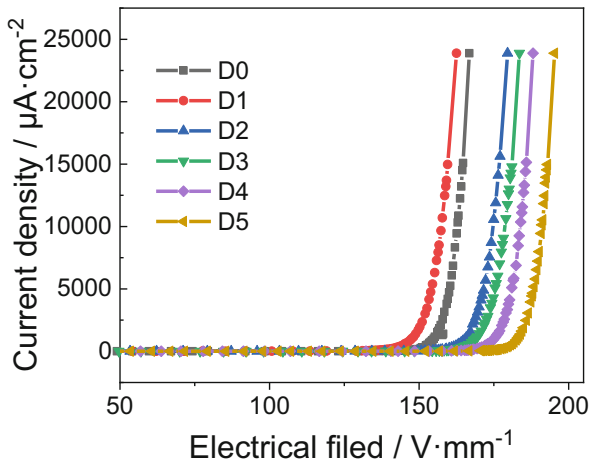


Fig. 4. Electrical field - current density curves of the ZnO varistors with different amounts of SnO₂ doping.

Figure 4 shows the E-J curves of ZnO varistors with different SnO₂ doping contents. The characteristic parameters extracted from the E-J curves are in agreement with those measured by the varistor parameter tester displayed in Table 1, which confirms the validity of the data.

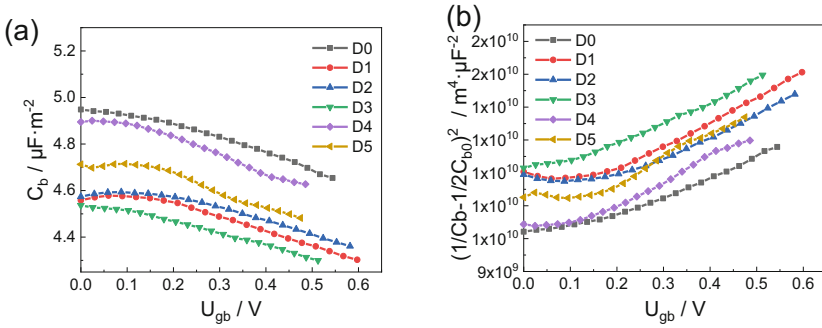


Fig. 5. C-V curves of ZnO varistors with different amounts of SnO₂ doping.

Table 2. Schottky barrier parameters of the ZnO varistors doped with different amounts of SnO₂ doping.

Sample	ϕ (eV)	N_d (10^{18} m^{-3})	N_s (10^{23} m^{-2})
D0	1.61	2.23	5.73
D1	1.33	1.81	4.72
D2	1.35	1.89	4.87
D3	1.33	1.85	5.16
D4	1.65	2.32	6.61
D5	1.77	2.32	7.09

Figure 5 and Table 2 show the C-V curves of the ZnO varistors and the Schottky barrier characteristic parameters calculated from the curves, respectively. As the SnO₂ content increases, the donor concentration N_d and the surface state density N_s decrease first and then increase, while the grain boundary barrier ϕ firstly decreases and then increases. The decrease of the donor concentration might be because SnO₂ promoted the formation of more liquid phase during sintering, allowing more additives to be incorporated into the Bi-rich phase instead of permeating to ZnO. The increase of N_d and N_s is because Sn⁴⁺ enters the ZnO lattice and becomes donor doping, increasing the donor concentration. Meanwhile, the oxygen produced by the decomposition of SnO₂ at high temperature increases the partial pressure of oxygen at the grain boundaries, resulting in more surface state density, according to the equation $\text{SnO}_2 \leftrightarrow 2\text{Sn}_{\text{Zn}} + 2e' + \text{O}_2$.

Figure 6 shows the 2 ms square-waveform energy withstanding capacity of the ZnO varistors with different amounts of SnO₂ doping. As the SnO₂ content increases, the maximum energy withstanding capacity first decreases, then increases to 250 A at sample D4, and then decreases to 200 A at D5. The weakening of the energy withstanding capacity from sample D0 to D1 is due to the uneven grain distribution and more cavities in the ceramic. The improvement of the energy withstanding capacity from D1 to D4 is due to the more uniform grain distribution. The decrease of the energy withstanding capacity from D4 to D5 may be due to the fact that the donor doping has been saturated,

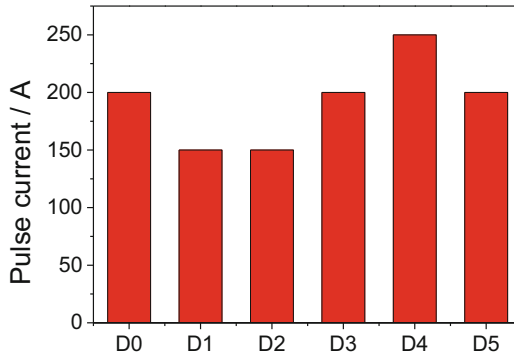


Fig. 6. 2 ms energy withstanding capacity of the ZnO varistors with different amounts of SnO₂ doping.

and the grain resistance cannot further decrease, while the number of grain boundaries further increases, which further aggravates the internal stress.

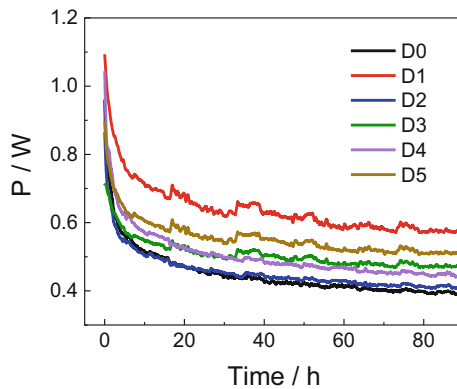


Fig. 7. Accelerated aging performance of the ZnO varistors with different amounts of SnO₂ doping.

In order to verify whether the ZnO varistors can run stably in the circuit, we tested the accelerated aging performance of the samples for 96 h. The accelerated aging curves are shown in Fig. 7, and the performance parameters are displayed in Table 3. The power consumption of D0 - D5 decreases first and then stabilizes as time goes on, and the K_{ct} of all the varistors is less than 1, indicating that all samples can run stably in the circuit. The power consumption is the product of the loading voltage and current of the sample, so the power consumption during aging is roughly proportional to the breakdown voltage gradient and the leakage current [4, 23]. D0 has the second smallest breakdown voltage gradient and the smallest leakage current, thus its aging power consumption is the least. D1 has the smallest breakdown voltage gradient, however, its leakage current is the largest, so its aging power consumption is the largest.

Table 3. Accelerated aging parameters of the ZnO varistors with different amounts of SnO₂ doping.

No	P _{1h} (W)	P _{96h} (W)	K _{ct}
D0	0.77	0.39	0.50
D1	0.97	0.58	0.60
D2	0.73	0.41	0.56
D3	0.69	0.48	0.70
D4	0.85	0.45	0.53
D5	0.77	0.50	0.65

4 Conclusions

This work systematically explored the influence of SnO₂ on the microstructure and the performance of the ZnO varistors. When the doping amount of SnO₂ was small, the main effect was to reduce the phase transition temperature of Bi₂O₃ and promote the ZnO grain growth. As the doping amount increased, SnO₂ led to the formation of the Zn₂SnO₄ spinel, inhibited the ZnO grain growth, promoted the uniform distribution of grains, and improved the uniformity of the structure of the ZnO varistors. During sintering process, Sn⁴⁺ entered the ZnO lattice, increasing the donor concentration. The oxygen produced by the decomposition of SnO₂ increased the partial pressure of oxygen at the grain boundaries, resulting in more surface state density. Sample D4 has the best comprehensive performance with the potential gradient of 208.5 V · mm⁻¹, the nonlinear coefficient of 43.42, the leakage current density of 1 μA, the 5 kA voltage ratio of 1.73, the energy withstanding capacity of 250 A suffering from 18 shocks of 2 ms square waveform pulse impacts, and the ageing coefficient K_{ct} of 0.53.

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